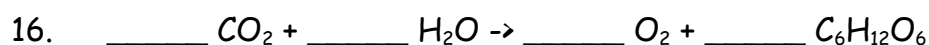
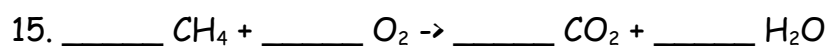
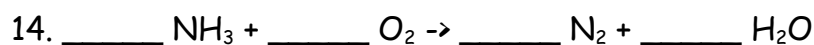
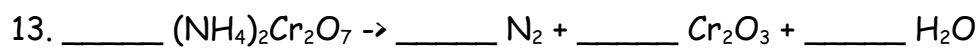
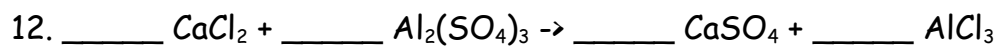
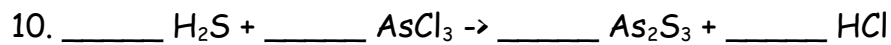
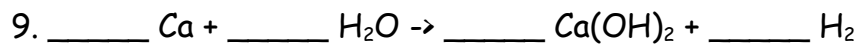
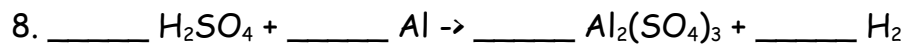
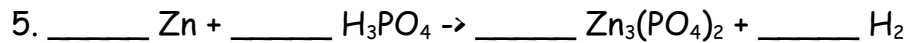
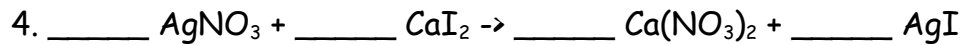
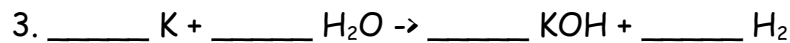
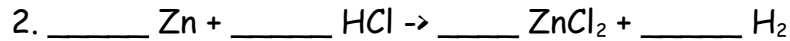
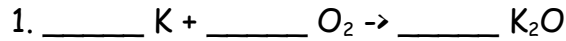


Name: \_\_\_\_\_

## Balancing Practice

Balance the following equations:



Rewrite each of the following word equations as balanced chemical equations:

1. sodium chloride + fluorine gas  $\rightarrow$  sodium fluoride + chlorine gas

2. iron (III) chloride + sodium hydroxide  $\rightarrow$  iron (III) hydroxide + sodium chloride

3. silver oxide  $\rightarrow$  silver + oxygen gas

4. phosphorus + oxygen gas  $\rightarrow$  diphosphorus pentoxide

5. potassium chlorate  $\rightarrow$  potassium chloride + oxygen gas

6. barium chloride + sodium sulfate  $\rightarrow$  barium sulfate + sodium chloride

7. hydrogen gas + nitrogen monoxide  $\rightarrow$  water + nitrogen gas

8. zinc + lead (II) nitrate  $\rightarrow$  zinc nitrate + lead

9. sodium phosphate + calcium chloride  $\rightarrow$  calcium phosphate + sodium chloride

10. copper + silver nitrate  $\rightarrow$  silver + copper (II) nitrate